

Ring Expansion and Rearrangement Reactions of *N*-Heteroarylmethyl Radicals

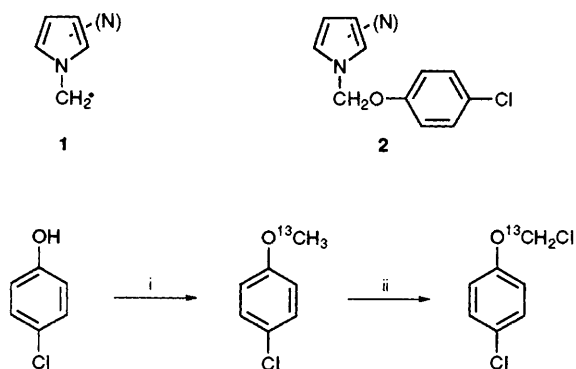
Jill F. McLellan, Hamish McNab* and Thomas W. Muir

Department of Chemistry, The University of Edinburgh, West Mains Road, Edinburgh, UK EH9 3JJ

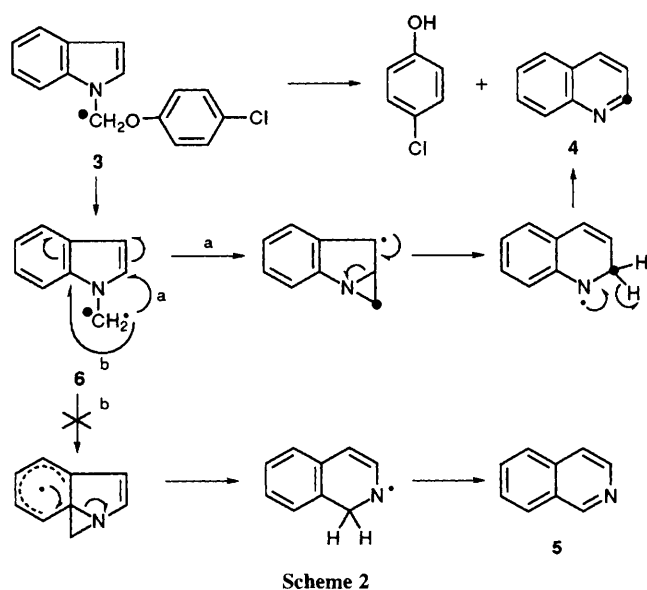
Flash vacuum pyrolysis of *N*-(*p*-chlorophenoxymethyl)azoles (e.g. **2** and **3**) gives ring expanded products (e.g. **4** and **8**) formed via *N*-heteroarylmethyl radicals (e.g. **1** and **6**); the mechanism of ring expansion generally involves a novel 'walk'-rearrangement of the radical centre.

There is considerable current interest in ring-expansion reactions of carbon-centred radicals, generated in solution under standard reductive conditions.¹ We describe here a gas-phase variant of these processes which leads to oxidative ring-expansion of *N*-heteroarylmethyl radicals **1** via novel 'walk' rearrangements of the intermediates.

The radicals **1** were generated (along with the *p*-chlorophenoxy radical) by flash vacuum pyrolysis (FVP) of the corresponding *N*-(4-chlorophenoxy)methyl derivative **2** at 700 °C. The precursors **2**† were themselves obtained in ca. 60% yield by alkylation of the parent heterocycle using α ,4-dichloroanisole under basic conditions in dimethyl sulfoxide solution.² ¹³C-Labelled dichloroanisole was obtained by the method shown in Scheme 1.^{3,4}



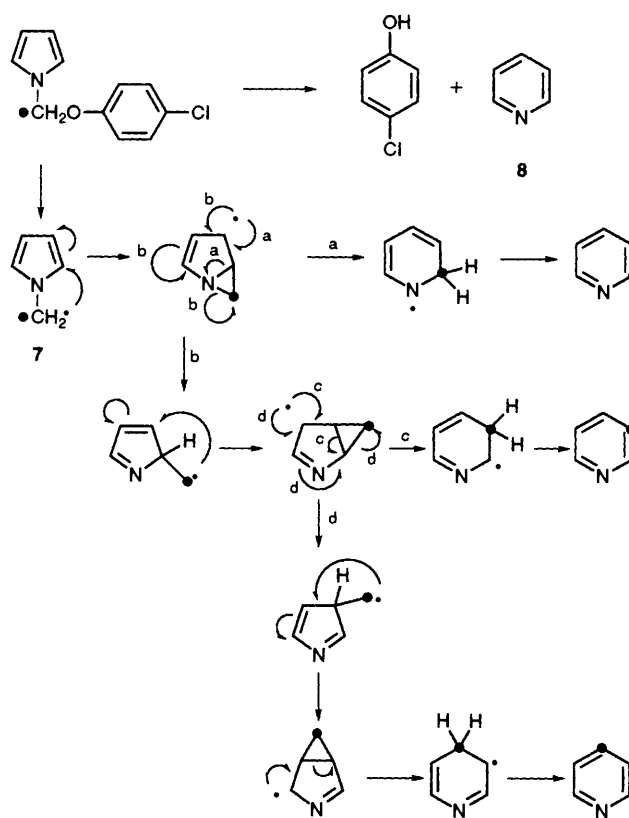
Scheme 1 Reagents and conditions: i, ¹³CH₃I, K₂CO₃, dimethylformamide; ii, PCl₅, heat



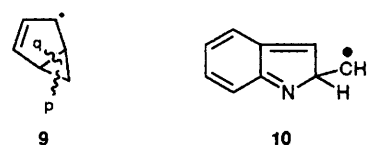
Scheme 2

FVP of the indolyl derivative **3** at 700–800 °C (10⁻² Torr‡) gives rise to *p*-chlorophenol (90%) as expected, and to quinoline **4**, (66%) as the only ring-expanded product. No isoquinoline **5** could be detected. The expansion process is therefore initiated exclusively by attack of the *N*-indolylmethyl radical **6** at the 2-position of the indole ring (Scheme 2 route a). Well-precedented attack at the 7a-position by a neophyl-type rearrangement^{5,6} (Scheme 2 route b) which would give rise to isoquinoline, clearly does not occur in this case. The mechanism was confirmed by pyrolysis of a ¹³C-labelled derivative (highlighted atoms in Scheme 2), and, as predicted, the majority of the label (>90%) is found at C-2 of the quinoline (¹³C NMR).

Although ring-expansion of the *N*-pyrrolylmethyl radical **7** similarly gives pyridine (**8**; 59%), the detailed mechanism of the rearrangement as revealed by the labelling experiment (¹³C NMR), differs significantly from the *N*-indolylmethyl

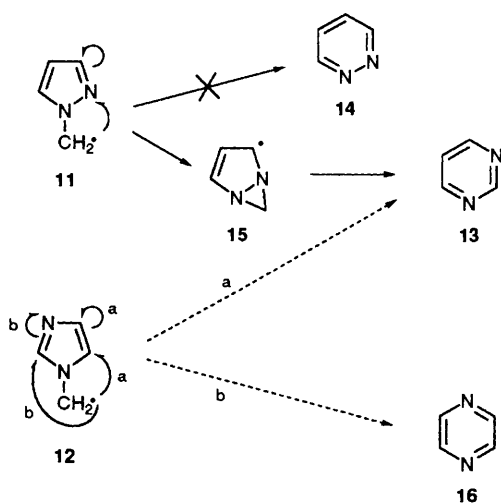


Scheme 3



† All new compounds were characterised by their spectra and by elemental analysis.

‡ 1 Torr ≈ 133 Pa.



Scheme 4

case. Surprisingly, the label is distributed over all positions of the pyridine ring, with the majority located at the 3-position. This result may be explained by a 'walk'-rearrangement of the initial radical (Scheme 3).§ The occurrence of such a rearrangement is consistent with the results of an EPR study of the hydrocarbon analogue **9**, in which exocyclic cleavage (bond p- leading to the 'walk') is known to be favoured over endocyclic cleavage (bond q- leading to ring expansion).⁸ In the pyrrolyl case, ring expansion by route c (Scheme 3) is apparently most favourable, and indeed it is known that drastic pyrolysis of *N*-alkylpyrroles under radical-chain conditions^{9,10} often gives 3-substituted pyridines as the major ring-expanded product.

§ A 1,5-shift of the *N*-substituent prior to radical formation could give a similar labelling pattern, but this is known to require much higher temperatures in our apparatus.⁷

A similar mechanism *via* the relatively disfavoured *ortho*-quinonoid intermediate **10** can explain residual label (<10%), found at the 3-position in the indolylmethyl experiment.

Ring expansion of the *N*-azolylmethyl radicals **11** and **12** produced contrasting results (Scheme 4). Thus the *N*-pyrazolylmethyl radical **11** expands regiospecifically to give pyrimidine **13** rather than pyridazine **14** and this is probably due to the cleavage of the particularly weak N-N bond of the diaziridinyl intermediate **15**. These results confirm that **11** is a viable intermediate in the pyrolysis of dipyrzolylmethane,¹¹ from which pyrimidine has also been obtained. In contrast, the *N*-imidazolylmethyl radical **12** yields both pyrimidine **13** and pyrazine **16** in 3.7:1.0 ratio.

We are grateful to the University of Edinburgh for a vacation scholarship (to T. W. M.).

Received, 22nd February 1993; Com. 3/01052J

References

- (a) P. Dowd and S.-C. Choi, *Tetrahedron Lett.*, 1989, **30**, 6129, and references therein; (b) J. E. Baldwin, R. M. Adlington and J. Robertson, *J. Chem. Soc., Chem. Commun.*, 1988, 1404; (c) C. W. Ellwood and G. Pattenden, *Tetrahedron Lett.*, 1991, **32**, 1591.
- H. Heaney and S. V. Ley, *J. Chem. Soc., Perkin Trans. 1*, 1973, 499.
- cf. J. I. G. Cadogan, C. L. Hickson and H. McNab, *J. Chem. Soc., Perkin Trans. 1*, 1985, 1885.
- H. Gross and W. Bürger, *Org. Synth. Coll. Vol. V*, 1973, 221.
- J. I. G. Cadogan, C. L. Hickson and H. McNab, *Tetrahedron*, 1986, **42**, 2135, and references therein.
- H. McNab, *J. Chem. Soc., Chem. Commun.*, 1990, 543.
- H. McNab and M. E.-A. Murray, *J. Chem. Soc., Perkin Trans. 1*, 1988, 333.
- (a) R. Sustmann and F. Lübke, *J. Am. Chem. Soc.*, 1976, **98**, 6037; (b) R. Sustmann and F. Lübke, *Chem. Ber.*, 1979, **112**, 42.
- J. M. Patterson, C. F. Mayer and W. T. Smith, Jr., *J. Org. Chem.*, 1975, **40**, 1511.
- A. Pictet, *Ber. Deut. Chem. Ges.*, 1905, **38**, 1946.
- J. D. Pérez, G. I. Yranzo, M. A. Ferraris, R. M. Claramunt, C. López and J. Elguero, *Tetrahedron*, 1988, **44**, 6429.